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The densities (kg/m³) of water, alcohol and the solution may be taken to be 998, 798 and 895 respectively at 293 K. Calculate the following: (a) The volume percent of ethanol in the solution at 293 K (b) The molarity (c) The molality Solution Basis: 100 kg of 60% (weight) alcoholwater solution (a) Volume of 100 kg of solution is mass 100 ==0.1117

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