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Calculate the mass
percentage of
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and carbon tetrachloride (CCl_4) if 22 g of benzene is dissolved in 122 g of carbon tetrachloride. Ans:
Mass of solution =
Mass of C_6H_6 +
Mass of CCl_4 = 22 g + 122 g = 144 g
Mass % of benzene
= $\frac{22}{144} \times 100$
= 15.28 %

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