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stoichiometric questions later in Chapter 3 as well as in Chapter 4 (section 4.6 on solution stoichiometry), Chapter 5 (stoichiometry of heat and Hess's Law), Chapter 10 (stoichiometry of gaseous reactions), Chapter 20 (section 20.9 on electrolysis).

[Chapter 3. Stoichiometry: Calculations with Chemical ...](#)

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balanced equation. How many moles of oxygen are produced from 12 moles of potassium chlorate? 3. Using the equation from problem 2, how many moles of oxygen are produced from 14 moles of potassium chlorate? 4. Two molecules of hydrogen react with one molecule of oxygen to produce two molecules of water.

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